

dissolved in Et<sub>2</sub>O, dried and the Et<sub>2</sub>O was removed to leave a crude solid which on CC over 50 g silica gel (60–120 mesh) furnished glut-5(10)-en-1-one, C<sub>30</sub>H<sub>48</sub>O (3), mp 312°, [ $\alpha$ ]<sub>D</sub> + 30° (CHCl<sub>3</sub>). It did not respond to a Zimmermann test; its UV, IR, <sup>1</sup>H NMR and MS data were similar to those in the lit. [3].

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### 3-ACETYLMASLINIC ACID FROM THE ROOT BARK OF *TERMINALIA ALATA*\*

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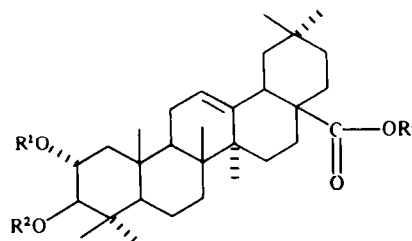
**Key Word Index**—*Terminalia alata*; Combretaceae; triterpenoids; 3-acetylmasic acid.

**Abstract**—A new triterpene acid, 3-acetylmasic acid, has been isolated from the root bark of *Terminalia alata* together with oleanolic acid, arjunic acid, arjunolic acid and arjunetin.

The isolation of triterpenoids from the heartwood of *Terminalia alata* Heyne ex Roth (syn. *T. tomentosa* W. & A.) was reported recently [1, 2]. Continuing our studies on the chemical constituents of the genus *Terminalia*, we report here the isolation of a new triterpene acid, identified as 3-acetylmasic acid, from the root bark of *T. alata* together with the known compounds oleanolic acid, arjunic acid, arjunolic acid and arjunetin.

Extraction of the ground root bark with CHCl<sub>3</sub> and EtOAc afforded a mixture of triterpenoids. Separation by repeated column chromatography and preparative TLC over silica gel led to the isolation of the above known triterpene acids and the new acid TARB-2. The compound TARB-2, mp 192–195°, [ $\alpha$ ]<sub>D</sub> + 32°, analysed for C<sub>32</sub>H<sub>50</sub>O<sub>5</sub> and gave a positive Liebermann–Burchard test and yellow colour with tetranitromethane. Its IR spectrum showed the presence of hydroxyl (3500 cm<sup>-1</sup>), ester carbonyl (1740 cm<sup>-1</sup>) and carboxyl (1690 cm<sup>-1</sup>) groups. The <sup>1</sup>H NMR spectrum showed the resonances for seven tertiary methyls, one acetate and olefinic groups. In addition, it showed the presence of CHOH with a signal at

$\delta$  3.29 (m) and a CHOAc signal at 4.68 (d,  $J$  = 12 Hz). The large coupling constant indicated a diaxial relation; therefore the hydroxyl and acetoxyl groups are in diequatorial orientation. Acetylation with acetic anhydride and pyridine gave diacetylmasic acid (1b), which on treatment with diazomethane afforded diacetylmethyl



**1** R<sup>1</sup> = R<sup>2</sup> = R<sup>3</sup> = H

**1a** R<sup>1</sup> = R<sup>3</sup> = H; R<sup>2</sup> = Ac

**1b** R<sup>1</sup> = R<sup>2</sup> = Ac; R<sup>3</sup> = H

**1c** R<sup>1</sup> = R<sup>2</sup> = Ac; R<sup>3</sup> = Me

maslinate (1c). Alkaline hydrolysis of TARB-2 gave maslinic acid (1) [2]. The above data lead to the identification of TARB-2 as 3-acetylmasiclinic acid (1a).

Although the occurrence of acetyloleanolic acid in plants is very common, this is the first report of the isolation of 3-acetylmasiclinic acid (1a) from a plant and, to our knowledge, it has also not been prepared from maslinic acid. The isolation of maslinic acid [1, 2], as well as its acetate, from *T. alata* is of biogenetic significance in providing the missing link for a dihydroxytriterpene amongst a mixture of mono- to tetrahydroxytriterpenoids found in the plant. The methanol extract gave a crude material which afforded a small amount of ellagic acid.

#### EXPERIMENTAL

Mps are uncorr. IR spectra were measured on KBr discs and  $^1\text{H}$  NMR spectra were recorded in  $\text{CDCl}_3$  soln using TMS as internal standard. The root bark of *T. alata* was obtained from CIMAP Experimental Station, Hebbale, Coorg District, Karnataka State, India and a voucher specimen has been deposited at CIMAP, Bangalore.

**Extraction and isolation.** Dried ground root bark (650 g) of *T. alata* was first defatted with hexane and extracted with  $\text{CHCl}_3$ , EtOAc and MeOH successively. The  $\text{CHCl}_3$  extract gave a material (2.0 g) which was separated into hexane (0.22 g) and  $\text{Et}_2\text{O}$  (1.1 g) soluble fractions. The hexane-soluble fraction on repeated crystallization from MeOH gave oleanolic acid (TARB-1; 80 mg), mp 310–311° [2]. The  $\text{Et}_2\text{O}$ -soluble fraction was found by TLC (silica gel;  $\text{CHCl}_3$ –MeOH, 98:2) to consist of one major and three minor compounds. CC over silica gel (100–200 mesh), eluting with  $\text{CHCl}_3$  and a  $\text{CHCl}_3$ –MeOH mixture of increasing polarity, gave a further amount of oleanolic acid (20 mg), a mixture of oleanolic acid and TARB-2 (60 mg),

arjunic acid (TARB-3; 625 mg), mp 334–335°, [3] and arjunolic acid (TARB-4; 50 mg), mp 330–332° [2]. The EtOAc extract gave a mixture (10 g) which was subjected to repeated CC and prep. TLC over silica gel: oleanolic acid (50 mg), TARB-2 (40 mg), arjunic acid (500 mg), arjunolic acid (60 mg) and arjunetin (TARB-5, 65 mg), mp 238–240° [4].

Compound TARB-2 (3-acetylmasiclinic acid, 1a) mp 192–195° (from  $\text{CHCl}_3$ –MeOH),  $[\alpha]_D^{25} + 32^\circ$  (c 0.5;  $\text{CHCl}_3$ ). IR  $\nu_{\text{max}}$   $\text{cm}^{-1}$ : 3500 (OH), 1740 (OAc), 1695 (COOH), 1460, 1265, 830 ( $-\text{C}=\text{CH}-$ ).  $^1\text{H}$  NMR:  $\delta$  0.70, 0.81, 0.85, 0.95, 1.05, 1.07 (6 s, 7 Me), 2.02 (s, OAc), 3.29 (m, 1H, H-2), 3.62 (br s, OH), 4.62 (d, 1H,  $J = 12$  Hz, H-3). Found: C, 74.76; H, 9.75. Calc. for  $\text{C}_{32}\text{H}_{50}\text{O}_5$ : C, 74.71; H, 9.72%. Acetate (Ac<sub>2</sub>O–pyridine), mp 232–234° (MeOH), identical to diacetylmasiclinic acid (1b) (mmp, IR,  $^1\text{H}$  NMR, co-TLC). Methyl ester diacetate mp 176–178° ( $\text{CH}_2\text{N}_2$ – $\text{Et}_2\text{O}$ ), identical to diacetylmethyl maslinate (1c) [1] (mmp, IR,  $^1\text{H}$  NMR, co-TLC). Hydrolysis of TARB-2 (15 mg) with 6% KOH in MeOH (10 ml) for 6 hr under reflux afforded after the usual work-up maslinic acid (1), mp 268–270° [2].

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